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Note

TWO COMPOUNDS FROM PEUCEDANUM DISSOLUTUM

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A new compound, 3'(R)-O- β -D-glucopyranosyl-3',4'-dihydroxanthyletin (1), and a known compound, prim-O-glucosylcimifugin (2), were isolated from the roots of *Peucedanum dissolutum*. The structure of 1 was elucidated by spectral evidence and chemical reaction. The NMR signals of carbons and protons of 2 were assigned for the first time by analysis of ¹H–¹H COSY, HMQC and HMBC spectra.

Keywords: Peucedanum dissolutum; 3'(R)-*O*- β -D-Glucopyranosyl-3',4'-dihydroxanthyletin; Prim-O-glucosyl-cimifugin

INTRODUCTION

Peucedanum dissolutum is a plant of Umbelliferae. In some areas of China the root of some plants of the *Peucedanum* genus, including *Peucedanum dissolutum*, have been used as Qianhu, a traditional Chinese medicine to cure diseases such as cough due to "pathogenic wind-heat, accumulation of phlegm and heat in the lung". So far there are no reports about the chemistry of the title plant. We have studied the chemical constituents of *Peucedanum dissolutum*. A new coumarin, along with a known chromone, was isolated and the structure of the new coumarin was elucidated as 3'(R)-O- β -D-glucopyranosyl-3',4'-dihydroxanthyletin by spectral analysis. The absolute configuration was deduced by chemical correlation with a known compound. This paper describes the isolation and structural elucidation of the two compounds.

RESULTS AND DISCUSSION

Compound 1 was obtained as colorless needles. The quasi-molecular ion peak $[M + 1]^+$ at m/z 409.1518 in the high-resolution FAB mass spectrum indicated the molecular formula to

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be $C_{20}H_{24}O_9$. The signals at 1705, 1622 cm⁻¹ in the IR spectrum were assigned to carbonyl and the aromatic system of a coumarin skeleton respectively. The ¹H NMR spectrum in the aromatic proton region of 1 revealed a pair of doublets at δ 6.45 (1H, d, J = 9.5 Hz), 8.05 (1H, d, J = 9.5 Hz) and a pair of singlets at δ 7.64 (1H, s), 7.34 (1H, s), which were attributed to the C₃-H, C₄-H signals of the α -pyrone ring system and the signals of C₅-H, C_8 -H of the benzene ring, indicating 1 to be a coumarin substituted at C-6 and C-7. In the ¹H NMR spectrum, the proton signals at δ 3.75 (1H, dd, J = 9.9, 3.2 Hz, H-3') and at δ 2.52 (1H, dd, J = 13.6, 9.9 Hz, H-4'), 3.61 (1H, dd, J = 13.6, 3.2 Hz, H-4') indicated that the substituted moiety between C-6 and C-7 formed a dihydropyran ring, and comparison with chemical shifts and the coupling pattern of the skeleton of smyrinol showed an attached group at C-3' [1]. In the HMQC spectrum (Fig. 1), the two proton signals at δ 2.52, 3.61 were correlated with the carbon signal at δ 33.54, showing no substituted group at C-4'; the same signals at δ 2.52, 3.61 were also correlated with the carbon signals at δ 131.51 (C-5) and 129.11 (C-6) in HMBC spectrum, confirming the above conclusion. The Molish reaction of 1 was positive and paper chromatography showed the existence of glucose. The signals at δ 102.84, 75.07, 78.63, 71.58, 77.93 and 62.71 in the ¹³C NMR spectrum were due to a glucose group. In the HMBC spectrum, the proton signal at δ 5.16 (glc-H-1) was correlated with the carbon signal at δ 79.33 (C-3'). This evidence indicates that the glucose is attached to C-3'.

To determine the absolute configuration of C-3', a chemical correlation with a known compound was carried out. On acid hydrolysis, **1** gave a product that was identified as 3'(R)-hydroxy-3',4'-dihydroxanthyletin (**3**) by comparison of its spectral data and optical rotation with those reported in the literature [2]; accordingly, the absolute configuration of C-3' in **1** was also established as *R* (Fig. 2). The chemical structure of **1** was finally elucidated as 3'(R)-O- β -D-glucopyranosyl-3',4'-dihydroxanthyletin. The signals in the ¹H and ¹³C NMR spectra were assigned by HMQC and HMBC spectra and are listed in the experimental section.

Compound 2 was isolated as a yellow oil, and gave a positive Molish reaction. ESI-MS gave a quasi-molecular ion peak at m/z 469. Combining the data of ¹H, ¹³C NMR and



FIGURE 1 Structures and key HMBC correlations of 1 and 2.



FIGURE 2 Acid hydrolysis of 1.

DEPT spectra, the molecular formula M/Z 2 was elucidated as $C_{22}H_{28}O_{11}$. The ¹H and ¹³C NMR spectra showed the characteristics of chromone. The DEPT spectrum showed three methyls, three methylenes, eight methines and eight quaternary carbons. The correlations in ¹H—¹H COSY, HMQC and HMBC spectra indicated that the C-6 and C-7 positions of chromone formed part of a dihydrofuran ring whose C-2' was attached to an isopropanol group, C-5 linked with a methyloxy group, and C-2 linked with a hydroxymethyl group that was attached to a glucose. The structure of **2** was identical with prim-O-glucosylcimifugin reported in the literature [3]. However, there are some wrong assignments of proton and carbon signals in the literature, and we have reassigned the proton and carbon signals of **2** by ¹H–¹H COSY, HMQC and HMBC spectra as listed in the Experimental section.

EXPERIMENTAL

General Experimental Procedures

Mps were determined on an X-4 micro melting-point apparatus, and uncorrected. UV spectra were recorded on a Shimadzu UV-2501 PC spectrophotometer in MeOH solution. IR spectra were obtained on a Nicolet Impact-410 spectrometer. 1D and 2D NMR spectra were recorded on a Bruker-DRX-400 spectrometer using TMS as internal standard. FABMS were measured on a MAT 212 mass spectrometer. ESIMS were taken on a PE-Mariner AFI-TOF mass spectrometer. Optical rotations were determined on a Perkin-Elmer 241 automatic polarimeter at 20°C. Silica gel H (10–40 μ m) was used for column chromatography.

Plant Material

Roots of *Peucedanum dissolutum* were collected in Chongqing city, China, in November 1998, and identified by Professor Xianqi Liu, Nanchuan Institute of Materia Medica in Chongqing, China. A voucher specimen (No 981103) has been deposited in the Department of Natural Medicinal Chemistry, China Pharmaceutical University.

Extraction and Isolation

The dried root (1.31 kg) was extracted $3 \times \text{with } 95\%$ ethanol (3 L) at 80°C . After filtration and evaporation, the residue (156 g) was dissolved in water (1.5 L) and the solution was extracted $3 \times \text{with}$ light petroleum $(60-90^{\circ}\text{C}, 1.5 \text{ L})$, EtOAc (1.5 L) and n-BuOH (1.5 L), respectively. The EtOAc extract (67 g) was subjected to column chromatography on silica gel (350 g) eluted with a mixture of CHCl₃ and MeOH of increasing polarity (each Fr. 300 ml). After evaporation of Frs. 15–28, eluted with CHCl₃–MeOH (90:10), the residue (2.6 g) was further separated by silica gel CC (100 g) eluted with CHCl₃–MeOH, and compound **1** X.-L. WU et al.

(23 mg) was obtained from the elution of $CHCl_3$ –MeOH (92:8) and **2** (39 mg) was obtained from the elution of $CHCl_3$ –MeOH (88:12).

3'(R)-O- β -D-Glucopyranosyl-3', 4'-dihydroxanthyletin (1)

Colorless cubic crystals, mp 206–207°C, $[\alpha]_D - 3.1$ (*c* 0.5, CHCl₃). UV λ_{max} (nm): 393.0, 325.5, 293.0; IR ν_{max} (cm⁻¹): 3490, 3371, 1705, 1622, 1269, 1117, 1054, 856; ¹H NMR (400 MHz, CDCl₃) δ (ppm): 6.45 (1H, d, J = 9.5 Hz, H-3), 8.05 (1H, d, J = 9.5 Hz, H-4), 7.64 (1H, s, H-5), 7.34 (1H, s, H-8), 3.75 (1H, dd, J = 9.9, 3.2 Hz, H-3'), 2.52 (1H, dd, J = 13.6, 9.9 Hz, H-4'), 3.61 (1H, dd, J = 13.6, 3.2 Hz, H-4'), 1.45 (3H, s, C-2'-CH₃), 1.43 (3H, s, C-2'-CH₃), 5.16 (1H, d, J = 7.4 Hz, H-1"), 3.93–3.69 (4H, m, H-2"-H-5"), 4.14 (1H, m, H-6"), 3.93 (1H, m, H-6"); ¹³C NMR (100 MHz, CDCl₃) δ (ppm): 163.79 (C-2), 114.35 (C-3), 146.10 (C-4), 131.51 (C-5), 129.11 (C-6), 160.32 (C-7), 104.07 (C-8), 155.68 (C-9), 114.98 (C-10), 74.53 (C-2'), 79.33 (C-3'), 33.54 (C-4'), 27.55 (C-2'-CH₃), 23.67 (C-2'-CH₃), 102.84 (C-1"), 75.07 (C-2"), 78.63 (C-3"), 71.58 (C-4"), 77.93 (C-5"), 62.71 (C-6"); HR-FABMS *m*/*z*: 409.1518 (C₂₀H₂₄O₉ calcd 409.1499 for [M + H]⁺).

Acid Hydrolysis of 3'(R)-O- β -D-Glucopyranosyl-3', 4'-dihydroxanthyletin (1)

Compound **1** (10 mg) dissolved in MeOH (3.0 mL) was added to 15% HCl (1.5 mL) and the reaction mixture refluxed for 1 h. The solution was then neutralized with 10% NaOH, and extracted with CHCl₃. The glucose in the water layer was identified by paper chromatography. The CHCl₃ layer was dried with Na₂SO₄, and evaporated, and the residue was purified with preparative TLC, developed with light petroleum–EtOAc (3:1) to give 3'(R)-hydroxy-3',4'-dihydroxanthyletin (**3**, 5 mg).

3'(R)-Hydroxy-3', 4'-dihydroxanthyletin (3)

White cubic crystals, mp 176.0–178.0°C, [α]_D – 13.9 (CHCl₃, *c* 0.05), literature [2]: –11.0 (CHCl₃). IR ν_{max} (cm⁻¹): 3384, 2975, 1706, 1628, 1575, 1386, 1145, 815; ¹H NMR (400 MHz, CDCl₃) δ (ppm): 6.19 (1H, d, J = 9.4 Hz, H-3), 7.91 (1H, d, J = 9.4 Hz, H-4), 7.44 (1H, s, H-5), 6.72 (1H, s, H-8), 3.48 (1H, dd, J = 10.2, 2.0 Hz, H-3'), 2.97 (1H, dd, J = 13.7, 2.0 Hz, H-4'), 2.34 (1H, dd, J = 13.7, 10.2 Hz, H-4'), 1.15 (3H, s, C-2'-CH₃), 1.12 (3H, s, C-2'-CH₃). ¹³C NMR (100 MHz, CDCl₃) δ (ppm): 160.65 (C-2), 111.18 (C-3), 144.73 (C-4), 130.34 (C-5), 125.11 (C-6), 159.51 (C-7), 101.55 (C-8), 153.65 (C-9), 110.92 (C-10), 76.74 (C-2'), 71.77 (C-3'), 31.34 (C-4'), 26.28 (C-2'-CH₃), 24.71 (C-2'-CH₃).

Prim-O-glucosylcimifugin (2)

Yellow oil, $[\alpha]_D + 6.3$ (*c* 0.6, CHCl₃). UV λ_{max} (nm): 293.0, 214.0; IR ν_{max} (cm⁻¹): 3353, 2924, 2853, 1731, 1657, 1608, 1461, 1382, 1115; ¹H NMR (400 MHz, CDCl₃): δ (ppm): 6.56 (1H, s, H-3), 6.79 (1H, s, H-8), 4.96 (1H, m, H-2'), 3.52 (2H, m, H-3'), 1.48 (3H, s, C-4'-CH₃), 1.43 (3H, s, C-4'-CH₃), 4.12 (3H, s, OCH₃), 4.79 (1H, d, *J* = 15.0 Hz, -CH₂--), 4.95 (1H, d, *J* = 15.0 Hz, -CH₂--), 4.62 (1H, d, *J* = 7.6 Hz, H-1"), 3.48-3.59 (4H, m, H-2"-H-5"), 4.07 (1H, m, H-6"), 3.87 (1H, m, H-6"); ¹³C NMR (100 MHz, CD₃OD): δ (ppm): 165.31 (C-2), 111.27 (C-3), 179.91 (C-4), 157.25 (C-5), 118.70 (C-6), 167.39 (C-7), 94.88 (C-8), 161.39 (C-9), 112.64 (C-10), 92.89 (C-2'), 29.06 (C-3'), 72.57 (C-4'), 25.68 (C-4'-CH₃), 25.58 (C-4'-CH₃), 61.37 (OCH₃), 67.60 (-CH₂-), 104.30 (C-1"), 75.22 (C-2"), 78.38 (C-3"), 71.78 (C-4"), 78.20 (C-5"), 62.94 (C-6"); ESIMS *m/z*: 469 [M + H]⁺.

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References

- Dzhafarov, Z.R., Kuliev, Z.A., Vdovin, A.D., Kuliev, A.A., Malikov, V.M. and Ismailov, N.M. (1992), Chem. Nat. Compd. 28, 27–31.
- [2] Lemmich, J. and Nielsen, B.E. (1969), Tetrahedron Lett. 1, 3–5.
- [3] Sasaki, H., Taguchi, H., Endo, T. and Yosioka, I. (1982), Chem. Pharm. Bull. 30, 3555-3562.